## **AMENDMENTS TO THE CLAIMS:**

Claims 1-6 (Canceled)

7. (Currently amended) A positive active material comprising:

a composite oxide which <u>comprises</u> is <u>constituted of at least</u> lithium (Li), manganese (Mn), nickel (Ni), cobalt (Co), and oxygen (O) and is represented by the following chemical composition formula:

 $\label{eq:linear_control_control} \begin{array}{ll} Li_aMn_bNi_cCo_dO_e & \text{(Chemical composition formula 1)} \\ [[()]]wherein \ 0< a \le 1.3 \\ |b\text{-}c| \le 0.05 \\ 0.6 \le d < 1 \\ 1.7 \le e \le 2.3 \\ b\text{+}c\text{+}d\text{=}1[[)]], \ \underline{and} \end{array}$ 

wherein said composite oxide comprises an oxide which is other than  $\underline{\text{LiMn}_{0.05}\text{Ni}_{0.05}\text{Co}_{0.9}\text{O}_2}$ , and  $b\neq 0.1$ .

8. (Currently amended) A positive active material comprising:

a composite oxide which <u>comprises</u> is constituted of at least lithium (Li), manganese (Mn), nickel (Ni), cobalt (Co), and oxygen (O) and is represented by the following chemical composition formula:

 $\label{eq:limits} \begin{array}{ll} Li_aMn_bNi_cCo_dO_e & \text{ (Chemical composition formula 1)} \\ [[()] wherein \ 0< a \le 1.3 \\ |b\text{-}c|<0.03 \\ 0.8 \le d < 1 \\ 1.7 \le e \le 2.3 \\ b\text{+}c\text{+}d\text{=}1[[)]], \ and \end{array}$ 

wherein said composite oxide comprises an oxide which is other than  $\underline{\text{LiMn}_{0.05}\text{Ni}_{0.05}\text{Co}_{0.9}\text{O}_{2}}$ , and  $b\neq 0.1$ .

- 9. (Currently amended) A non-aqueous electrolyte battery, comprising: having
  a positive electrode including containing the positive active material of claim 7; [[,]]
  a negative electrode; [[,]] and
  a non-aqueous electrolyte.
- 10. (Currently amended) A non-aqueous electrolyte battery, comprising: having a positive electrode including containing the positive active material of claim 8; [[,]] a negative electrode; [[,]] and a non-aqueous electrolyte.
- 11. (Currently amended) A non-aqueous electrolyte battery, comprising: having
  a positive electrode, a negative electrode, and a non-aqueous electrolyte, characterized in
  that

wherein the positive electrode comprises contains a lithium-manganese oxide (A) having a spinel structure and represented by the general formula  $\text{LiMn}_2\text{O}_4$  and a lithium-nickel-manganese-cobalt composite oxide (B) having an  $\alpha$ -NaFeO<sub>2</sub> type layer structure and represented by the general formula  $\text{Li}_a\text{Mn}_b\text{Ni}_c\text{Co}_d\text{O}_e$ ,

wherein a weight ratio of (A) to (B) is in a range from 5:95 to 10:90, and wherein

 $0 \le a \le 1.3$ 

 $|b-c| \le 0.05$ 

 $0.6 \le d \le 1$ 

 $1.7 \le e \le 2.3$ 

b+c+d=1.

12. (Currently amended) A non-aqueous electrolyte battery, <u>comprising</u>: having a positive electrode, a negative electrode, and a non-aqueous electrolyte, <del>characterized in that</del>

wherein the positive electrode comprises contains a lithium-manganese oxide (A) having a spinel structure and represented by the general formula  $\text{LiMn}_2\text{O}_4$  and a lithium-nickel-manganese-cobalt composite oxide (B) having an  $\alpha$ -NaFeO<sub>2</sub> type layer structure and represented by the general formula  $\text{Li}_a\text{Mn}_b\text{Ni}_c\text{Co}_d\text{O}_e$ ,

wherein a weight ratio of (A) to (B) is in a range from 5:95 to 10:90, and wherein

 $0 \le a \le 1.3$ 

|b-c| < 0.03

 $0.8 \le d \le 1$ 

 $1.7 \le e \le 2.3$ 

b+c+d=1.

- 13. (Currently amended) The non-aqueous electrolyte battery of claim 19 11, wherein eharacterized in that the positive electrode includes contains the (A) and the (B) in a proportion (weight ratio ration) of from 5:95 to 90:10.
- 14. (Currently amended) The non-aqueous electrolyte battery of claim 20 12, wherein eharacterized in that the positive electrode includes contains the (A) and the (B) in a proportion (weight ratio ration) of from 5:95 to 90:10.
- 15. (New) A positive active material comprising:

a composite oxide which comprises lithium (Li), manganese (Mn), nickel (Ni), cobalt (Co), and oxygen (O) and is represented by the following chemical composition formula:

Li<sub>a</sub>Mn<sub>b</sub>Ni<sub>c</sub>Co<sub>d</sub>O<sub>e</sub> (Chemical composition formula 1)

wherein  $0 \le a \le 1.3$ 

 $|b-c| \le 0.05$   $0.6 \le d \le 0.833$   $1.7 \le e \le 2.3$ b+c+d=1, and

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b≠0.1.

16. (New) A positive active material comprising:

a composite oxide which comprises lithium (Li), manganese (Mn), nickel (Ni), cobalt (Co), and oxygen (O) and is represented by the following chemical composition formula:

$$Li_aMn_bNi_cCo_dO_e$$
 (Chemical composition formula 1)

wherein  $0 \le a \le 1.3$ 

|b-c| < 0.03

 $0.8 \le d \le 0.833$ 

 $1.7 \le e \le 2.3$ 

b+c+d=1, and

b≠0.1.

- 17. (New) A non-aqueous electrolyte battery, comprising:
  - a positive electrode including the positive active material of claim 15;
  - a negative electrode; and
  - a non-aqueous electrolyte.
- 18. (Currently amended) A non-aqueous electrolyte battery, comprising:
  - a positive electrode including the positive active material of claim 16;
  - a negative electrode; and
  - a non-aqueous electrolyte.
- 19. (New) A non-aqueous electrolyte battery, comprising:

a positive electrode, a negative electrode, and a non-aqueous electrolyte,

wherein the positive electrode comprises a lithium-manganese oxide (A) having a spinel structure and represented by the general formula  $\text{LiMn}_2\text{O}_4$  and a lithium-nickel-manganese-cobalt composite oxide (B) having an  $\alpha$ -NaFeO<sub>2</sub> layer structure and represented by the general formula  $\text{Li}_a\text{Mn}_b\text{Ni}_c\text{Co}_d\text{O}_e$ ,

## wherein

 $0 \le a \le 1.3$ 

|b-c|≤0.05

0.9≤d<1

 $1.7 \le e \le 2.3$ 

b+c+d=1

b<0.05.

## 20. (New) A non-aqueous electrolyte battery, comprising:

a positive electrode, a negative electrode, and a non-aqueous electrolyte,

wherein the positive electrode comprises a lithium-manganese oxide (A) having a spinel structure and represented by the general formula  $\text{LiMn}_2\text{O}_4$  and a lithium-nickel-manganese-cobalt composite oxide (B) having an  $\alpha$ -NaFeO<sub>2</sub> layer structure and represented by the general formula  $\text{Li}_a\text{Mn}_b\text{Ni}_c\text{Co}_d\text{O}_e$ ,

## wherein

 $0 \le a \le 1.3$ 

|b-c| < 0.03

 $0.9 \le d \le 1$ 

 $1.7 \le e \le 2.3$ 

b+c+d=1

b<0.05.